

C/60

Government College, Daman

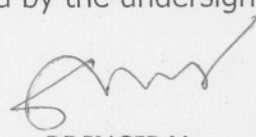
Ref:-GC/DMN/CHEM/PUR/WDS/2964

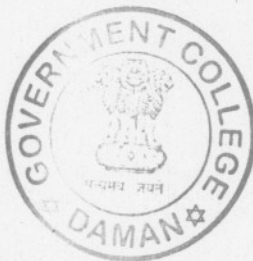
Date: 23/02/2012

Quotation Notice

Quotation duly sealed and super scribed as "Quotation for **Chemistry Lab.** are invited up to 1:00 p.m. on 05/03/2012 which shall be opened as far as possible on the same day at 3:00 pm. For the supply of stores mentioned in the schedule enclosed on the following terms and conditions.

1. The rates quoted should be strictly F.O.R. destination. Quotation suppliers may quote the rates for F. O. R. Vapi (W. Rly) only and shall have to bear the transport charges from Vapi to government College Daman. No extra charge for packing forwarding and insurance etc., shall be paid on the rates quoted. Quotations received, which do not confirm shall be summarily rejected.
2. The rates quoted should remain valid and operative for a period of 180 days from the date of opening of Quotation.
3. Only V.A.T as applicable shall be paid on the rates quoted.
4. **The rates should be quoted only for the items specified in the schedule and should be for the item of given specifications/mark/manufacturer.** As far as possible illustrated leaflets/catalogue should be enclosed with the Quotations. Where specification/mark/manufacturer is not specified the rates should be quoted only for standard quality. The decision of the undersigned for equivalent of specifications/ mark/ manufacturer shall be final.
5. Samples of the stores asked for shall have to be produced within stipulated time falling, which the Quotation for the said items shall be out rightly rejected.
6. The supplied of stores of inferior quality/standard or of different specifications other than that ordered/ specified and / or incomplete or broken articles will not be accepted. The supplier will have to replace the same and collect back the rejected articles at his own cost and risk. In the event of any dispute the decision of the undersigned shall be final and binding on suppliers.
7. Neither any advance payment shall be made nor shall condition for supply against through bank be accepted.
8. Quotations received which do not confirm with the terms and condition will be summarily rejected.
9. The item should have a warranty period as specified by the manufacturer with free installation in the laboratory.
10. Right to accept the Quotation in full or in part or to reject any or all the Quotation without assigning any reason there of is reserved by the undersigned.


PRINCIPAL,
GOVERNMENT COLLEGE,
DAMAN.



To
M/s _____

Government College, Dibrugarh

CHIEF GOVERNMENT COLLEGE, DIBRUGARH

DATE: 1/1/80

Question Paper

1. Question duly sealed and superimposed as "Question for Government College, Dibrugarh" and kept in the safe of the Controller of Examinations, Dibrugarh. The question paper shall be opened only on the day of the examination and shall be distributed to the candidates at the examination centre.

2. The rates quoted should be strictly F.O.B. Dibrugarh. Question paper rates should be quoted for F.O.B. Dibrugarh, W. B. and shall be valid for the entire duration of the examination. The rates should be quoted for the entire duration of the examination and shall be valid for the entire duration of the examination.

3. The rates quoted should be strictly F.O.B. Dibrugarh. Question paper rates should be quoted for F.O.B. Dibrugarh, W. B. and shall be valid for the entire duration of the examination.

4. The rates should be quoted only for the rates specified in the question paper and should be for the entire duration of the examination.

5. As far as possible, the rates should be quoted for the entire duration of the examination and shall be valid for the entire duration of the examination.

6. Question paper rates should be quoted only for the entire duration of the examination and shall be valid for the entire duration of the examination.

7. Samples of the rates should be quoted for the entire duration of the examination and shall be valid for the entire duration of the examination.

8. The rates should be quoted only for the entire duration of the examination and shall be valid for the entire duration of the examination.

9. The rates should be quoted only for the entire duration of the examination and shall be valid for the entire duration of the examination.

10. The rates should be quoted only for the entire duration of the examination and shall be valid for the entire duration of the examination.

11. The rates should be quoted only for the entire duration of the examination and shall be valid for the entire duration of the examination.

12. The rates should be quoted only for the entire duration of the examination and shall be valid for the entire duration of the examination.

13. The rates should be quoted only for the entire duration of the examination and shall be valid for the entire duration of the examination.

14. The rates should be quoted only for the entire duration of the examination and shall be valid for the entire duration of the examination.

15. The rates should be quoted only for the entire duration of the examination and shall be valid for the entire duration of the examination.




C/59

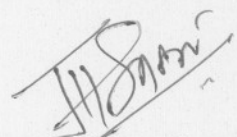
GOVERNMENT COLLEGE, DAMAN

CHEMISTRY DEPARTMENT

Description, specification and quantity of Water Distillation System with
Distillation Power Apparatus.

Description	Quantity	Specification
WATER DISTILLATION SYSTEM Distillation Units-Mono Quartz, panel Mounted with Quartz Condenser, 5ltr/hour, PH 6-7, pyrogen free, with distillation power supply apparatus, BOROSIL.	01 Nos. with distillation power supply apparatus	Borosil


Laboratory Assistant,
Chemistry Department,
Govt. College,
Daman.


Lecturer in Chemistry,
Govt. College,
Daman.

GOVERNMENT COLLEGE TAIWAN

CHEMISTRY DEPARTMENT

Examination questions and answers of 80th National Science Olympiad

Question for 1st year students

Question	Answer
1. A mixture of 100 g of 10% NaOH solution and 100 g of 10% HCl solution is allowed to react. The resulting solution is then diluted to 200 g. What is the pH of the resulting solution?	7.0
2. A 100 g sample of a mixture of NaCl and NaNO ₃ is dissolved in water. The solution is then treated with AgNO ₃ solution. The resulting precipitate is dried and weighed, giving a mass of 143.5 g. What is the mass percentage of NaCl in the original mixture?	58.5%
3. A 100 g sample of a mixture of NaCl and NaNO ₃ is dissolved in water. The solution is then treated with AgNO ₃ solution. The resulting precipitate is dried and weighed, giving a mass of 143.5 g. What is the mass percentage of NaCl in the original mixture?	58.5%
4. A 100 g sample of a mixture of NaCl and NaNO ₃ is dissolved in water. The solution is then treated with AgNO ₃ solution. The resulting precipitate is dried and weighed, giving a mass of 143.5 g. What is the mass percentage of NaCl in the original mixture?	58.5%
5. A 100 g sample of a mixture of NaCl and NaNO ₃ is dissolved in water. The solution is then treated with AgNO ₃ solution. The resulting precipitate is dried and weighed, giving a mass of 143.5 g. What is the mass percentage of NaCl in the original mixture?	58.5%
6. A 100 g sample of a mixture of NaCl and NaNO ₃ is dissolved in water. The solution is then treated with AgNO ₃ solution. The resulting precipitate is dried and weighed, giving a mass of 143.5 g. What is the mass percentage of NaCl in the original mixture?	58.5%
7. A 100 g sample of a mixture of NaCl and NaNO ₃ is dissolved in water. The solution is then treated with AgNO ₃ solution. The resulting precipitate is dried and weighed, giving a mass of 143.5 g. What is the mass percentage of NaCl in the original mixture?	58.5%
8. A 100 g sample of a mixture of NaCl and NaNO ₃ is dissolved in water. The solution is then treated with AgNO ₃ solution. The resulting precipitate is dried and weighed, giving a mass of 143.5 g. What is the mass percentage of NaCl in the original mixture?	58.5%
9. A 100 g sample of a mixture of NaCl and NaNO ₃ is dissolved in water. The solution is then treated with AgNO ₃ solution. The resulting precipitate is dried and weighed, giving a mass of 143.5 g. What is the mass percentage of NaCl in the original mixture?	58.5%
10. A 100 g sample of a mixture of NaCl and NaNO ₃ is dissolved in water. The solution is then treated with AgNO ₃ solution. The resulting precipitate is dried and weighed, giving a mass of 143.5 g. What is the mass percentage of NaCl in the original mixture?	58.5%

11/1/2000

Department of Chemistry
Government College
Taipei

Department of Chemistry
Government College
Taipei